

Table 3.--Chemical composition of 12 samples of oil shale from central Brooks Range, northern Alaska, and samples of the Green River, Chattanooga, and Pierre and Bearpaw Shales for comparison.

(Analysts and methods: SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub>, L. D. Cuss, colorimetric; Fe<sub>2</sub>O<sub>3</sub>, MgO, CaO, MnO, John Mensik & Claude Huffman, atomic absorption; Na<sub>2</sub>O and K<sub>2</sub>O, Wayne Mountjoy, internal standard flame photometer; TiO<sub>2</sub> and L.O.I., G. T. Burrow, colorimetric and gravimetric; Li and SO<sub>3</sub>, L. F. Rader, gravimetric; As, Se, and V, G. T. Burrow, colorimetric, nephelometric, and colorimetric; Ba and Pb, Wayne Mountjoy, colorimetric and atomic absorption; Cr, Mn, Sr, and Zn, Claude Huffman & John Mensik, atomic absorption; Co and Mo, L. F. Rader, colorimetric; Ni, E. J. Fennelly, colorimetric; eU, H. H. Kipp, alpha-beta count; Hg, Edward Knobell, atomic absorption spectroscopy; Ag, Margaret Hinkle, colorimetric field method; Au, H. W. Luki, H. M. Nakajima, & O. H. Van Sickle, colorimetric field method; Te, J. B. McHugh, colorimetric field method; specific gravity determinations (average of two measurements), R. F. Gantier. Major elements reported in percent; minor elements reported in ppm except where otherwise indicated).

Locality Number Field No. Laboratory No.	Alaska Oil Shale												Green River Shale/ K62-350X D115183	Chatt- anooga Shale/ LC201 D11245	Pierre Shale/ S57-37-1 D11505	Barpaw Shale/ C81-168-1 298401
	11 DL15215	12 DL15216	13 DL15217	14 DL15218	15 DL15219	16 DL15220	17 DL15221	18 DL15222	19 DL15223	20 DL15224	21 DL15225	22 DL15226				
SiO <sub>2</sub>	13.1	59.1	1.22	15.4	42.1	50.1	15.7	30.4	58.0	41.3	44.0	48.1	31.6	46.1	63.0	67.0
Al <sub>2</sub> O <sub>3</sub>	3.03	5.70	1.25	1.1	1.44	1.74	2.47	7.04	8.34	2.31	11.6	14.0	7.36	0.88	13.2	13.5
Total iron as Fe <sub>2</sub> O <sub>3</sub>	3.96	8.60	1.42	0.31	0.77	1.77	3.01	2.28	1.29	2.55	2.30	2.40	2.69	13.9	2.91	5.17
MgO	0.28	0.88	0.17	0.07	0.07	0.17	0.63	1.58	0.60	0.10	1.20	0.90	5.00	0.86	1.08	1.10
CaO	0.75	2.45	0.11	0.06	0.06	0.18	23.1	4.84	0.12	0.14	2.10	1.15	12.8	0.75	0.36	0.29
Na <sub>2</sub> O	0.28	0.62	0.16	0.50	0.71	0.20	0.17	0.75	0.63	0.97	1.14	1.29	2.24	0.44	0.49	0.70
K <sub>2</sub> O	0.61	1.13	0.43	0.28	0.22	0.28	0.45	0.98	1.70	0.48	2.23	2.60	2.02	2.95	3.45	2.53
TiO <sub>2</sub>	0.22	0.42	0.13	0.12	0.14	0.10	0.19	0.47	0.36	0.23	0.53	0.59	0.28	0.64	0.73	0.64
P <sub>2</sub> O <sub>5</sub>	0.15	0.55	0.17	0.03	<0.03	0.04	0.55	2.10	0.20	0.05	0.40	0.30	0.22	0.63	0.17	0.20
MnO	0.009	0.13	0.003	0.001	0.002	0.01	0.68	0.45	0.005	0.009	0.067	0.009	0.038	0.016	0.006	0.008
BaO	0.07	0.03	0.03	0.30	1.71	0.07	1.81	3.52	0.89	0.26	7.56	0.99	0.10	0.03	0.10	0.07
Total Sulfur as SO <sub>3</sub>	13.0	13.6	9.21	2.72	2.87	3.25	7.89	7.97	2.62	2.72	6.27	4.81	1.45	25.2	1.10	0.20
LOI at 1000°C	76.1	37.2	86.4	80.4	49.0	43.4	43.6	40.2	24.7	49.2	20.2	24.8	32.0	24.8	12.4	7.3
Total Y	111.6	110.5	103.8	101.4	100.1	100.5	100.3	102.6	99.5	100.3	99.0	101.1	97.5	100.6	92.0	88.7
As	49	200	37	63	67	110	37	110	27	110	24	29	43	240	43	17
B	52	80	53	32	35	57	110	180	60	250	270	140	270	240	240	240
Cr	40	110	24	18	14	16	19	250	260	48	280	430	250	300	150	100
Cu	40	320	38	8	2	25	15	480	32	34	65	100	47	88	66	27
Co	9	150	2	8	5	14	1	27	2	12	3	3	60	240	5	15
Pb	30	80	10	<10	<10	<10	20	10	10	<10	<10	<10	45	50	25	45
Mo	37	220	110	110	170	450	11	100	36	440	23	55	26	280	14	4
Ni	72	450	22	22	17	2	81	400	120	28	95	93	960	400	10	21
Se	<1	10	5	8	4	5	3	200	100	20	50	65	2	1	5	15
Sr	70	100	70	80	490	42	580	600	200	110	860	200	660	75	130	110
V	450	660	240	200	70	140	90	1700	1200	210	3100	1200	70	23	39	20
Zn	52	810	38	48	13	38	420	7000	76	34	140	280	57	250	42	97
eU	50	30	20	20	<10	<10	30	70	20	20	30	20	10	80	30	20
Ag (ppb)	3000	1400	1400	1200	3000	1600	900	1500	1000	650	750	800	-	-	-	-
As (ppb)	150	300	200	200	100	200	300	4000	400	300	300	300	-	-	-	-
Au (ppb)	150	150	<30	50	70	100	70	50	70	70	30	100	-	-	-	-
Te (ppb)	<50	<50	<50	<50	<50	<50	<50	<50	<50	<50	100	100	-	-	-	-
Sp. G.	1.3	1.0	1.2	1.1	1.5	1.2	1.0	1.8	1.9	1.7	2.1	2.1	-	-	-	-

1/ Mathematical total. No adjustments, such as oxygen included as Fe<sub>2</sub>O<sub>3</sub> if iron is pyrite, or inclusion of sulfur twice, due to measuring as SO<sub>3</sub> and also as LOI, have been made.

2/ Average mine-run shale from the Bureau of Mines Demonstration Mine near Rifle, Colorado. See Stanfield and others (1951, "composite sample", p. 4, tables 6, 15-20) for data on properties and composition of a sample of similar but not identical shale.

3/ Adit, about 1 mi. SW. on old Tennessee route 26 (now a boat landing road) from point where it joins Route 26 at top of descent to E. end of Sligo Bridge. See Bates and Strahil (1957) for data on mineralogy of shale from the same locality.

4/ Pierre Shale, Sharon Springs Member, NW<sup>1/4</sup> sec. 17, T. 93 N., R. 56 W., Yankton Co., S. Dak.

5/ Bearpaw Shale, 290 ft above base, NW<sup>1/4</sup> NW<sup>1/4</sup> sec. 3, T. 36 N., R. 8 W., Glacier Co., Mont.